

Prediction of Oral Absorption: 6-Fluoroquinolones



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*Strategies for Oral Drug Delivery
Oral Absorption Regulation and Evaluation
March 11-16, 2001 Garmisch Partenkirchen, Germany*

March, 2001

1

Prediction of Oral Absorption: 6-Fluoroquinolones

“Bioavailability prediction in drug development:
fluoroquinolones.” CICYT (SAF 96-1710)

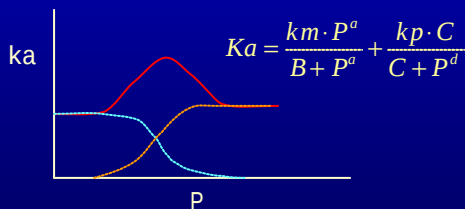
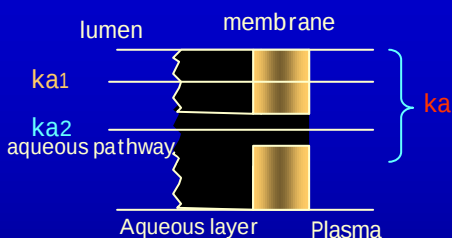
Director: Prof. JM Plá Delfina

- Absorption-partition relationships
- Bioavailability predictions

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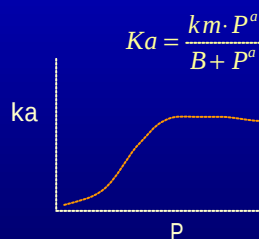
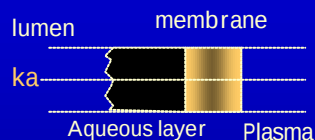
Absorption-Partition Correlations



$$Ka = \frac{km \cdot P^a}{B + P^a} + \frac{kp \cdot C}{C + P^d}$$

J. Pharmacokin. Biopharm 14(6),615-33 (1986)

J. Pharmacokin. Biopharm 15(6),633-43 (1987)

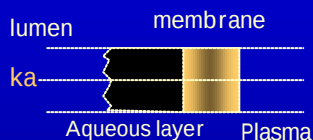
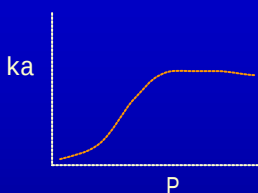


$$Ka = \frac{km \cdot P^a}{B + P^a}$$

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Absorption-Partition Correlations



Higuchi - Ho

$$\frac{1}{ka} = \frac{1}{kaq} + \frac{1}{kmem}$$

$$Ka = \frac{C \cdot P^d}{1 + E \cdot \sqrt{M} \cdot P^d}$$

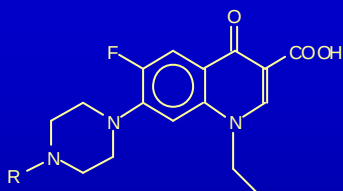
- Correlations in Colon or
- in Small intestine with Compounds MW>250

Journal of Pharmaceutical Sciences **84**, 777-782 (1995).

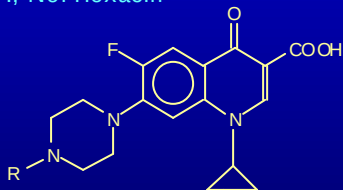
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Absorption-Partition Correlations



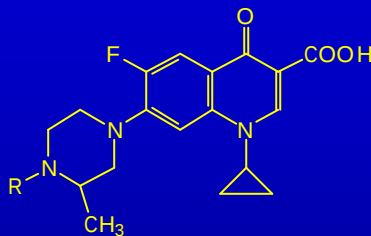
R= H, Norfloxacin



R= H, Cyprofloxacin

R= CH₃ , CH₃-(CH₂)_n - n=1-5

Homologous Compounds



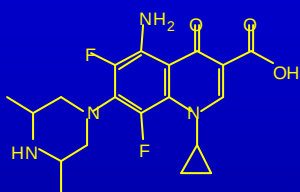
R	Name
H	CNV 97100
-CH ₃	CNV 97101
-CH ₂ -CH ₃	CNV 97102
-(CH ₂) ₂ -CH ₃	CNV 97103
-(CH ₂) ₃ -CH ₃	CNV 97104

CENAVISA S.A. Spain

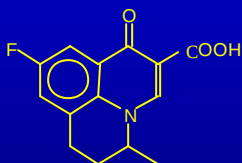
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Absorption-Partition Correlations



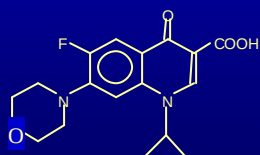
Sparfloxacin



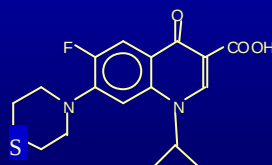
Flumequine



Sarafloxacin



CENAVISA 8804

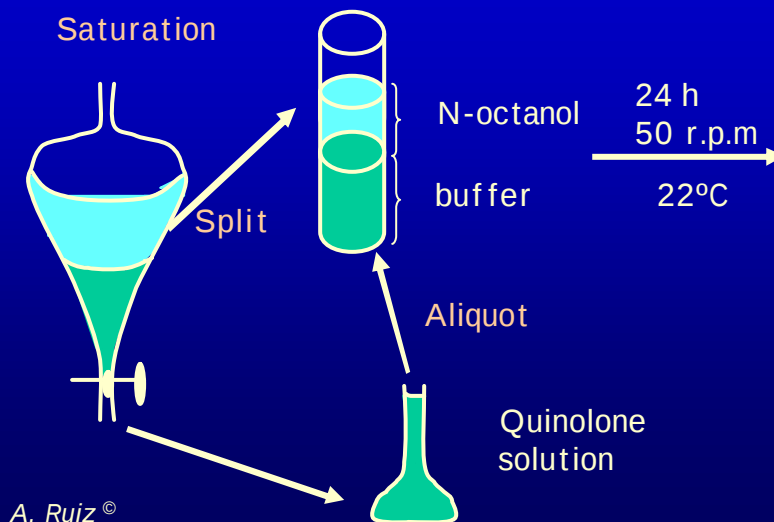


CENAVISA 8919

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Absorption-Partition Correlations



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Absorption-Partition Correlations

Partition Coefficient:

Organic Phase: n-Octanol

Aqueous Phase: aqueous buffer solution at pH. 7.00

Phase Volume adjusted for each compound

- Saturation
- Dissolution of quinolone in aqueous phase
- Equilibrating during 24 h at 22°C
- Analysis of the aqueous concentration by HPLC

$$P = \frac{C_o}{C_a} = \frac{(Q_{ai} - Q_{af})/V_o}{Q_{af}/V_a}$$

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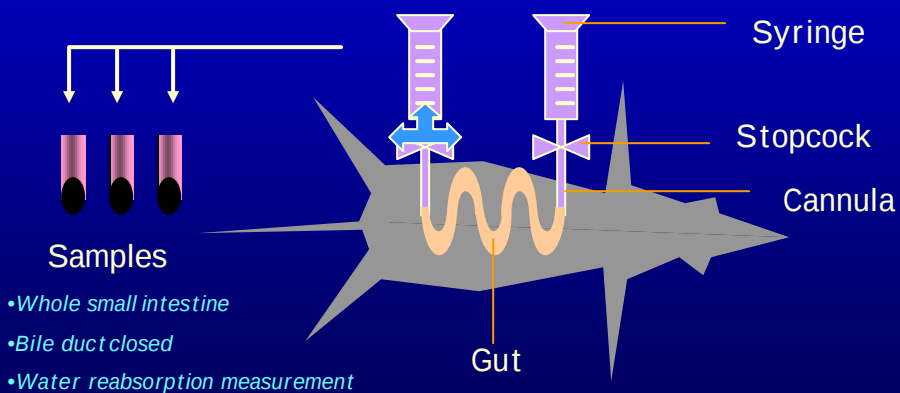
Absorption-Partition Correlations

Absorption studies:

- Solutions of compounds on saline buffered at pH. 7.00
- Concentration of each compound far enough of its C_s
- Perfused Volume: 10 mL

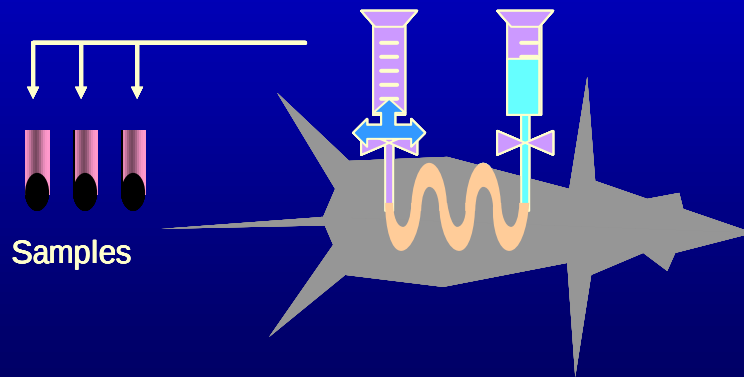
Absorption-Partition Correlations

Closed Loop Perfusion Technique: Doluisio's Method



Absorption-Partition Correlations

Closed Loop Perfusion Technique: Doluisio's Method

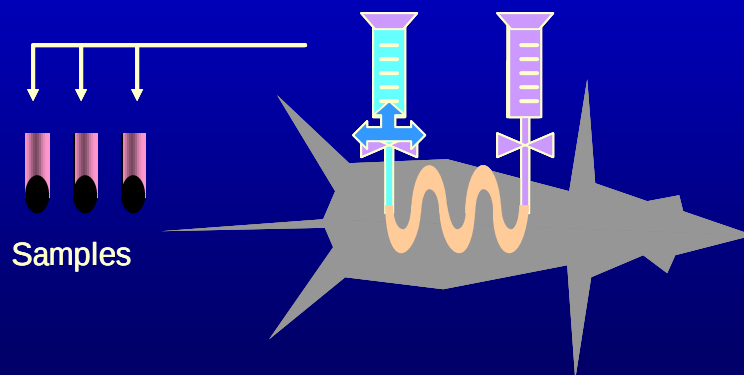


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Absorption-Partition Correlations

Closed Loop Perfusion Technique: Doluisio's Method

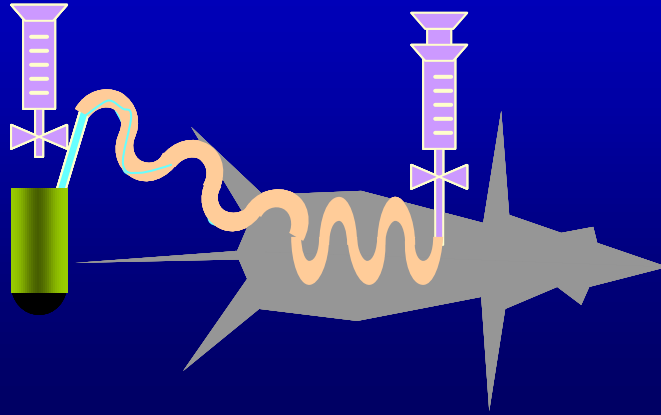


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Absorption-Partition Correlations

Closed Loop Perfusion Technique: Doluisio's Method



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Absorption-Partition Correlations

Water reabsorption correction

$$V_t = V_0 - k \cdot t$$

$$A_c = A_{\text{exp}} \cdot \frac{V_t}{V_0}$$

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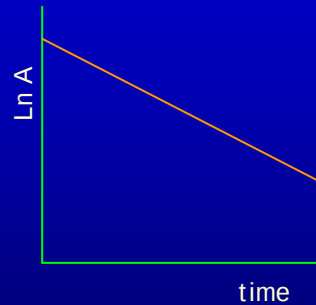
14

Absorption-Partition Correlations

Absorption rate constant:

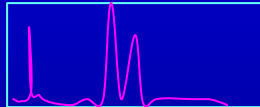
$$A = A_0 \cdot e^{-ka \cdot t}$$

k_a , mean value of six animals

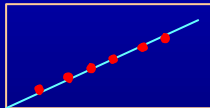


Absorption-Partition Correlations

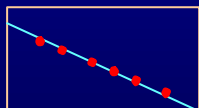
Analysis: HPLC fluorimetric detection



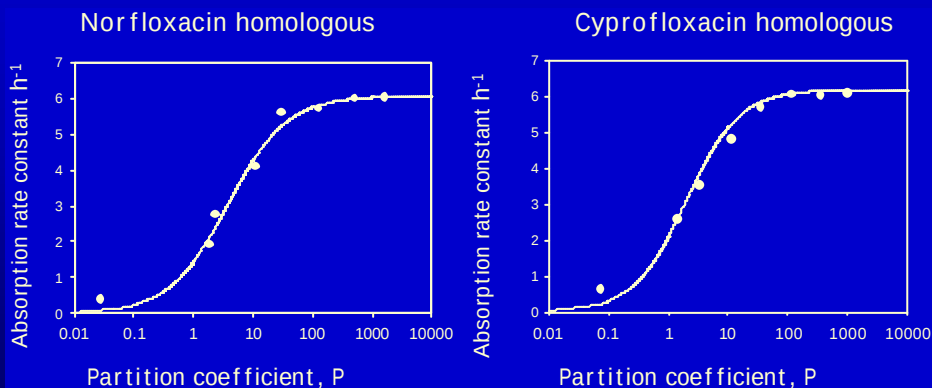
Method validation



Stability of compound in luminal fluid



Absorption-Partition Correlations



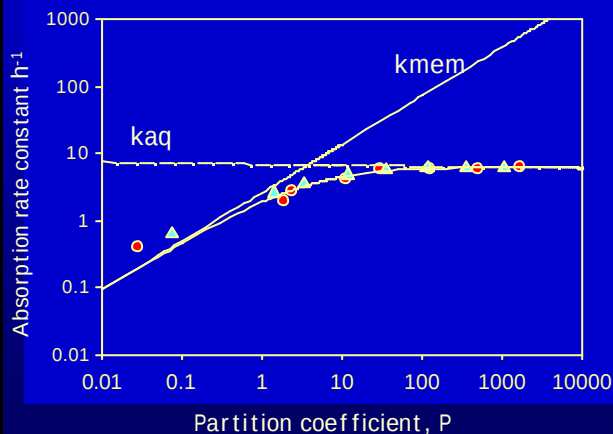
Journal of Pharmaceutical Sciences **84**, 777-782 (1995).

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Absorption-Partition Correlations

$$\frac{1}{ka} = \frac{1}{kaq} + \frac{1}{kmem}$$



Kaq (h ⁻¹)	Kmem (h ⁻¹)
7.073	0.195
6.944	0.397
6.923	3.322
6.801	4.070
6.782	4.795
6.667	6.210
6.649	14.813
6.541	15.359
6.524	30.267
6.422	33.761
6.405	79.701
6.309	85.387
6.293	181.075
6.201	233.991
6.186	389.164
6.099	545.016

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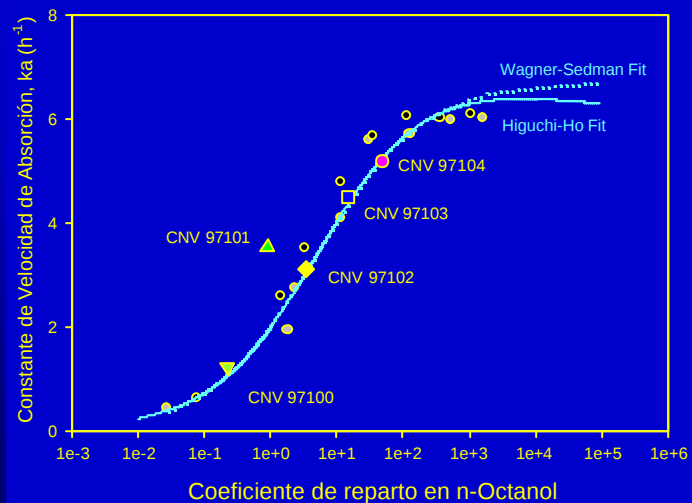
Absorption-Partition Correlations

Compound	MIC ₉₀	Compound	MIC ₉₀
Norfloxacin	0.1	Ciprofloxacin	0.025
N'-Methylnorfloxacin	0.1	N'-Methylciprofloxacin	0.025
N'-Ethylnorfloxacin	0.2	N'-Ethylciprofloxacin	0.05
N'-Propylnorfloxacin	0.4	N'-Propylciprofloxacin	0.1
N'-Butylnorfloxacin	0.4	N'-Butylciprofloxacin	0.2
N'-Pentylnorfloxacin	0.4	N'-Pentylciprofloxacin	0.2
N'-Hexylnorfloxacin	1.6	N'-Hexylciprofloxacin	0.4
N'-Heptylnorfloxacin	3.2	N'-Heptylciprofloxacin	0.4

MIC₉₀ of the compounds against *E. Coli* ATCC 25922

Journal of Pharmaceutical Sciences **84**, 777-782 (1995).

Absorption-Partition Correlations



Ana Ruiz-García. Ph.D Dissertation. University of Valencia. November 2000

Absorption-Partition Correlations

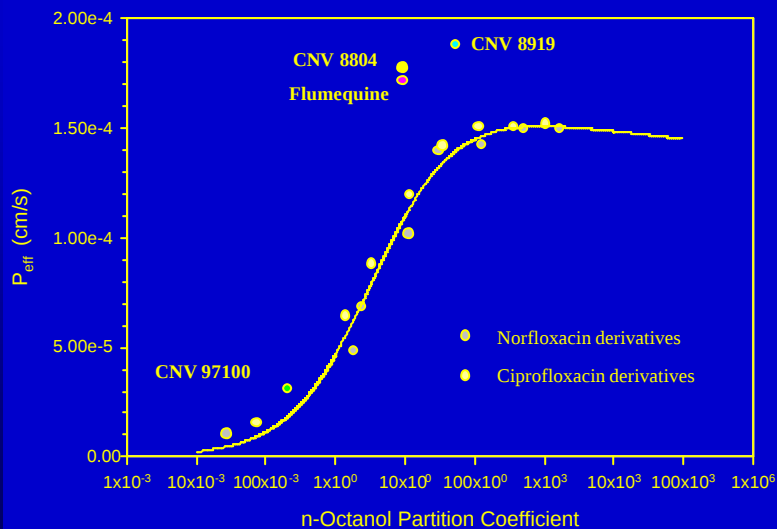
Quinolone	ka-pred HH	ka exp.	ER%
97100	1.06	1.24	-14.9
97101	1.91	3.54	-46.0
97102	3.06	3.12	-2.1
97103	4.35	4.5	-3.2
97104	5.20	5.16	0.8

Ana Ruiz-García. Ph.D Dissertation. University of Valencia. November 2000

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Absorption-Partition Correlations



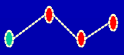
Journal of Pharmaceutical Sciences (4) 398-405, 1998.

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22

Absorption-Partition Correlations : QSAR methodologies in drug absorption

Instituto de Topología y Dinámica de Sistemas ITODYS.
Université Paris VII. France

1	<chem>CH3-COOH</chem>		0	1	0	0	0
2	<chem>CH3-CH2-CH2-CH3</chem>		1	0	1	0	1
3	<chem>CH3-CH(CH3)-CH3</chem>		1	0	1	1	0
Trazo			Vector				
			A1	A2	B11	B12	C1

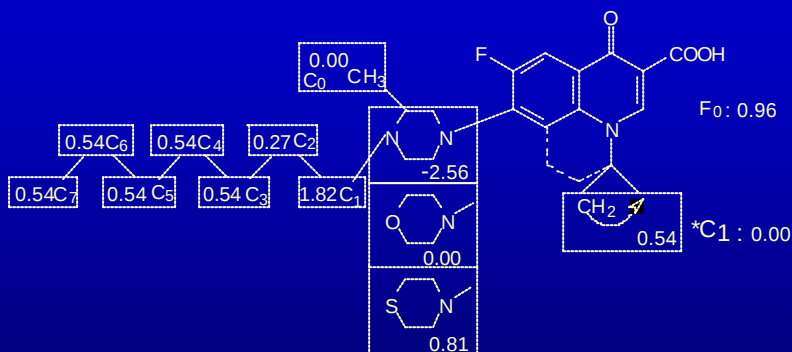
Vizet P. PATQSAR®. Version 2.55. 1997

Journal of Pharmaceutical Sciences (4) 398-405, 1998.

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23

Absorption-Partition Correlations : QSAR methodologies in drug absorption



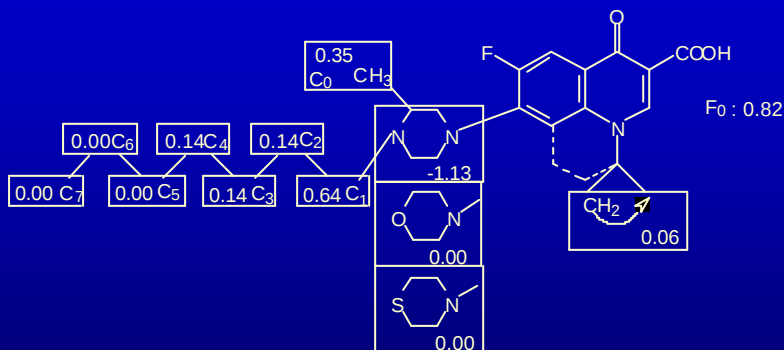
Lipophilicity-structure correlation

Journal of Pharmaceutical Sciences (4) 398-405, 1998.

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Absorption-Partition Correlations : QSAR methodologies in drug absorption



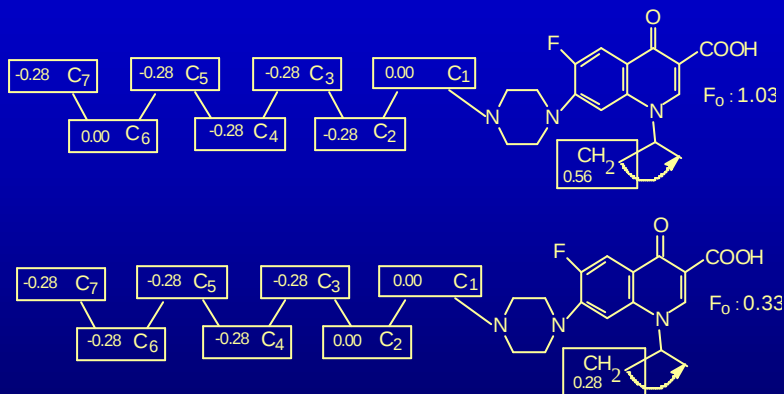
Absorption-structure correlation

Journal of Pharmaceutical Sciences (4) 398-405, 1998.

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Absorption-Partition Correlations : QSAR methodologies in drug absorption



Structure- antibacterial activity correlation

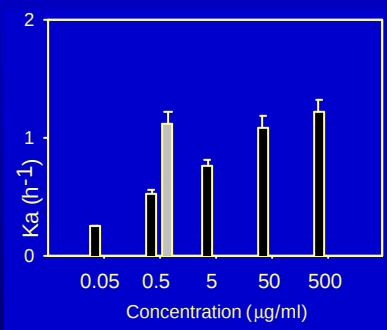
Journal of Pharmaceutical Sciences (4) 398-405, 1998.

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Absorption-Partition Correlations

CNV 97 100



Conc. (mg/mL)	k_a (h^{-1})	DE
500	1.215	0.107
50	1.081	0.110
5	0.757	0.050
0.5	0.523	0.040
0.05	0.242	0.010
0.5+V	1.114	0.113

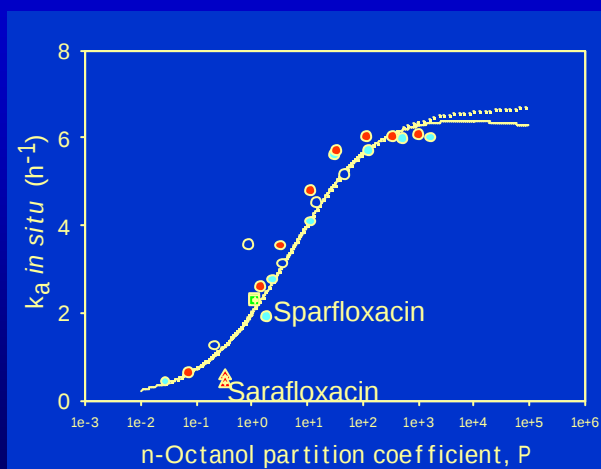
Conc (µg/mL)	500	50	5	0.5	0.05	0.5+V
500	-	-	-	-	-	-
50	NS	-	-	-	-	-
5	S	S	-	-	-	-
0.5	S	S	S	-	-	-
0.05	S	S	S	S	-	-
0.5+V	NS	NS	S	S	S	-

Gonzalez I. et al. V SEFIG meeting, Valencia 2001

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Absorption-Partition Correlations



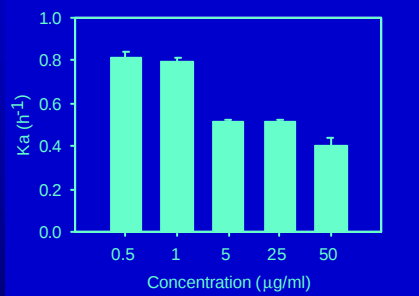
Fernández C. et al. V SEFIG meeting, Valencia 2001

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Absorption-Partition Correlations

Sarafloxacin



Conc(mg/mL)	$k_a(h^{-1})$	DE
50	0.399	0.044
25	0.512	0.016
5	0.513	0.014
1	0.789	0.025
0.5	0.816	0.026

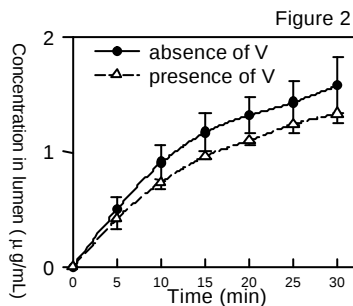
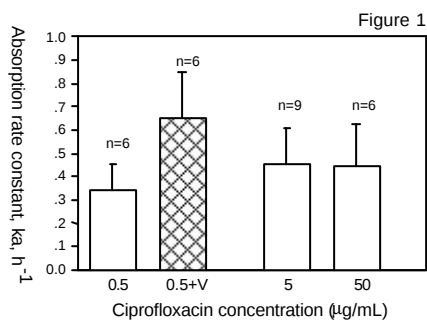
Conc (µg/mL)	50	25	5	1	0.5
50	-	-	-	-	-
25	S	-	-	-	-
5	S	NS	-	-	-
1	S	S	S	-	-
0.5	S	S	S	NS	-

Fernández C. et al. V SEFIG meeting, Valencia 2001

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STUDY OF THE INTESTINAL ABSORPTION OF CIPROFLOXACIN IN RATS.

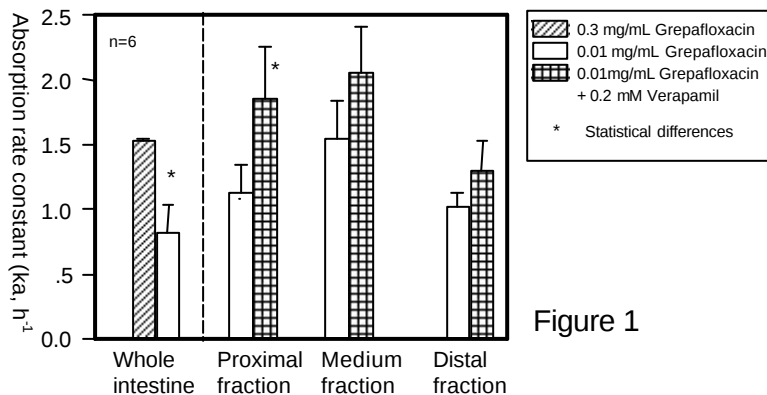


Rodríguez M et al. 1999 AAPS annual meeting

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INFLUENCE OF P-GLYCOPROTEIN ON THE INTESTINAL ABSORPTION OF GREPAFLOXACIN.



Sánchez-Castaño G. et al. 1999 AAPS annual meeting

Grepafloxacin donated by
GLAXO

Bioavailability prediction in drug development

$$F = Fa \cdot (1 - E_g) \cdot (1 - E_h)$$

E_g = gut or liver first pass effects

E_h = biliary excretion

Bioavailability prediction in drug development

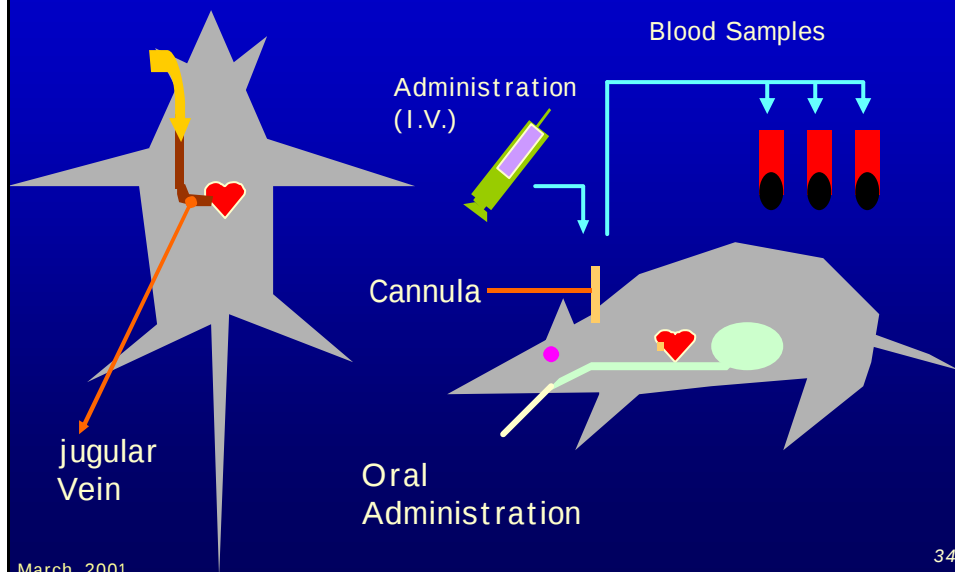
$$F_a = 1 - e^{-ka \cdot T}$$

$$F_a = 1 - e^{-\left(\frac{B \cdot P^a}{1 + C \cdot \sqrt{M} + P^a} \right) T}$$

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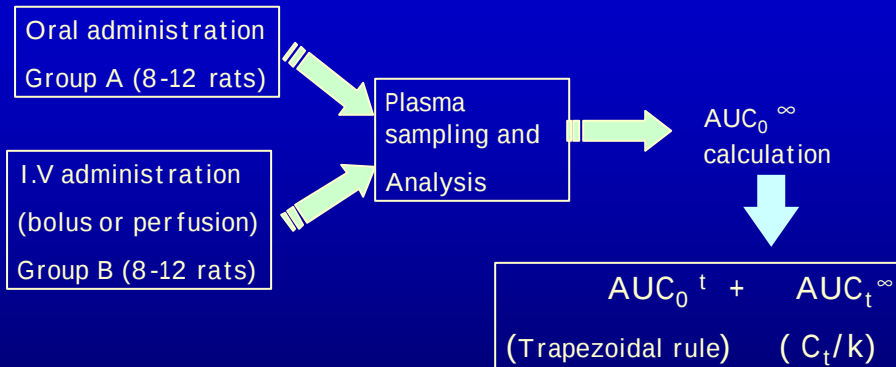
Bioavailability prediction in drug development



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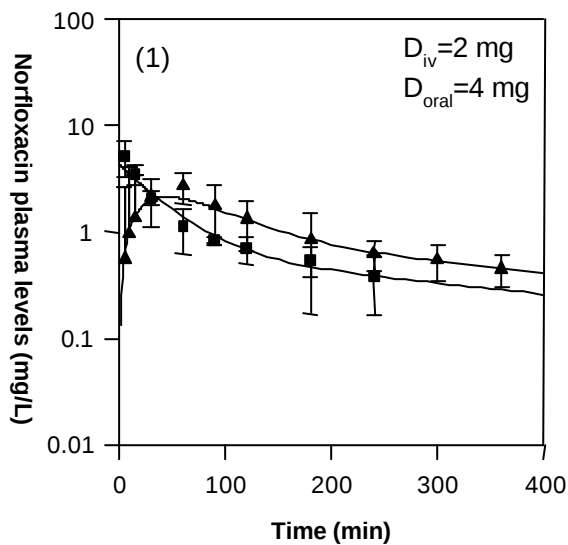
34

Bioavailability prediction in drug development



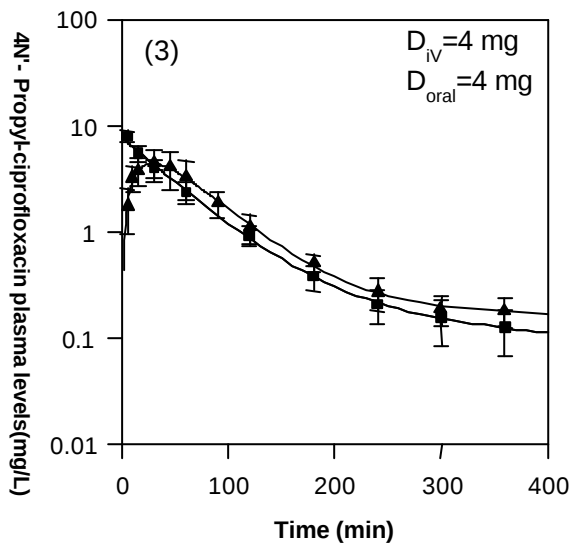
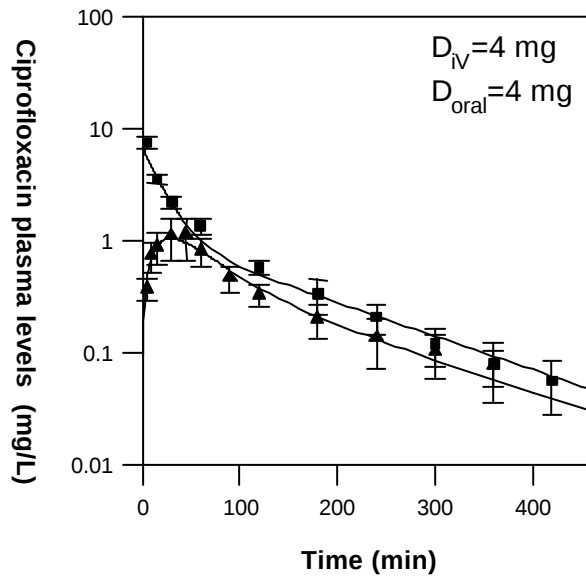
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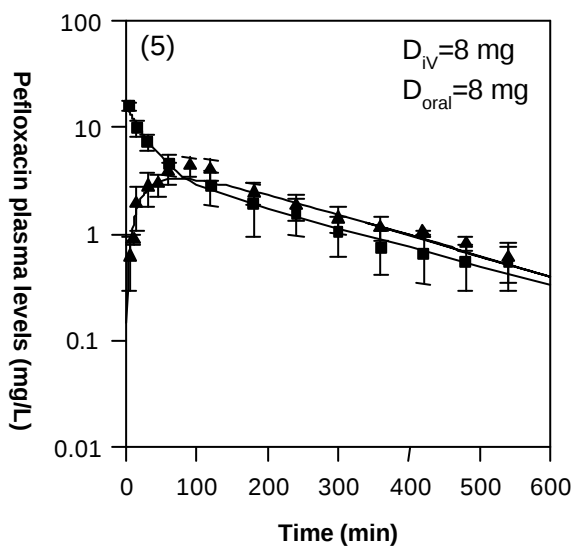
35



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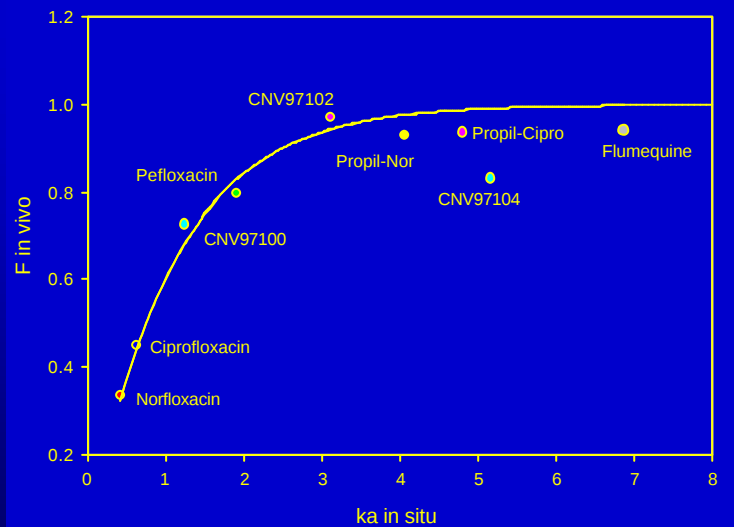
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Tested quinolone	Substituents in the formula			Molecular Weight (M)	P (n-octanol)	$k_a \text{ (h}^{-1}\text{) in situ}$
	R ₁	R ₂	R ₃			
Norfloxacin	Ethyl	H	H	319	0.028±0.002	0.42±0.07
Pefloxacin	Ethyl	Methyl	H	333	1.883±0.050	1.92±0.18
4N'-Propyl-norfloxacin	Ethyl	Propyl	H	361	11.27±0.27	4.07±0.40
Ciprofloxacin	Cyclopropyl	H	H	331	0.075±0.003	0.63±0.08
4N'-Propyl-ciprofloxacin	Cyclopropyl	Propyl	H	373	11.80±0.20	4.80±0.52
3'-Methyl-ciprofloxacin	Cyclopropyl	H	Methyl	345	0.209±0.001	1.24±0.07
Flumequine	Occupied by a 1-8 ring	Without 7N-piperazinyl substituent		261	9.40±0.90	6.87±1.09

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Journal of Pharmaceutical Sciences. 89(11)1395-1403(2000)

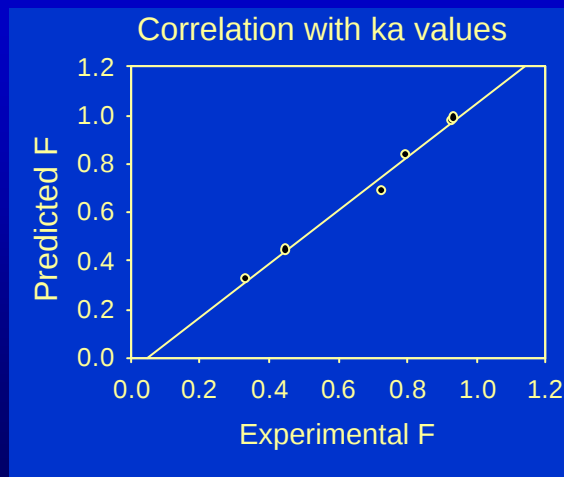
European Journal of Biopharmaceutics and Pharmacokinetics. 48,253-258 (1999).

Ana Ruiz-García. Ph.D Dissertation. University of Valencia. November 2000

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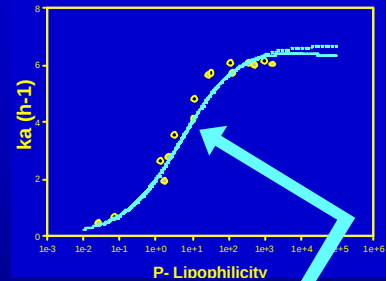
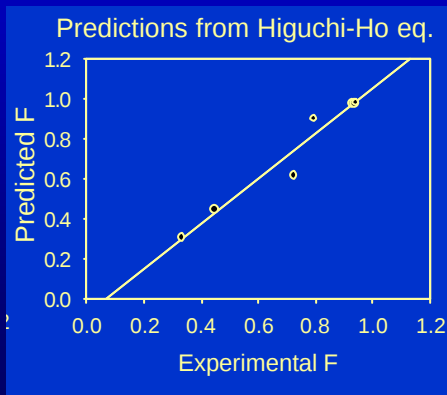
Bioavailability prediction in drug development



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Bioavailability prediction in drug development



$$F_a = 1 - e^{-\left(\frac{B \cdot P^a}{1 + C \cdot \sqrt{M} + P^a}\right) T}$$

Conclusions

1. Lipophilicity seems to be the main factor governing quinolone absorption
2. A Partition coefficient in n-Octanol over 10 produces an absorption constant $\sim 2 \text{ h}^{-1}$ $\longrightarrow$ $P_{eff} \sim 1 \cdot 10^{-4} \text{ cm/s}$
3. Quinolones with a in situ permeability over $2 \cdot 10^{-4} \text{ cm/s}$ have a bioavailability approx. 90%.

Conclusions

4. The absorption-partition relationship based on Higuchi-Ho equation has a good predictive performance at least with homologous quinolones; for heterologues compounds, the model is still able to predict the partition coefficient for maximum absorption.
5. The absorption rate constant k_a determined in situ in rat is a good predictive parameter of oral bioavailability for the studied compounds.
6. The efflux processes observed in situ and in vitro have not a significant influence in vivo in rats at the oral doses used.

Work in progress

1. In vitro Permeability studies (Caco-2):
 - Double Log-linear relationship P_{eff} in situ- P_{eff} in vitro.
 - Efflux process for some quinolones.
2. In situ absorption experiments of 97100 and 97101 at different concentrations: apparent active mechanism?
3. In vivo relevance of 97100 efflux.
4. In vitro metabolism of 97103.

Research Team

Chairman: José M. Plá-Delfina. Ph. D.

Professors

- M^a del Val Bermejo Sanz. Ph.D.
- Teresa Garrigues Pelufo. Ph.D.
- Virginia Merino Sanjuán. Ph.D.

Teaching assistants

- Gloria Sánchez Castaño Ph.D
- Ana Ruíz García Ph.D

Ph.D candidates

Carlos Fernández

Isabel González

Ricardo Nalda

Margarita Rodríguez