Spectral Assignments and Reference Data

Assignment of ¹H and ¹³C NMR data for (-)-methyl thyrsiflorin A and some scopadulan precursors

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Received 28 December 2000; Accepted 8 February 2001

The ¹H and ¹³C NMR spectral analysis of synthetic (-)methylthyrsiflorin A and 10 scopadulan precursors is reported. Resonance assignments were based on oneand two-dimensional NMR techniques, which included ¹H, ¹³C, DEPT and HMQC and also 1D NOE difference spectroscopy. Copyright © 2001 John Wiley & Sons, Ltd.

KEYWORDS: NMR; ¹H NMR; ¹³C NMR; scopadulan diterpenes; scopadulan precursors

INTRODUCTION

A number of structurally unique tetracyclic terpenoids have been isolated from the medicinal plants *Scoparia dulcis* L. and *Calceolaria thyrsiflora* (both belonging to the Scrophulariaceae family).¹ These compounds and some semisynthetic analogues have been shown to be potentially useful for treating disorders such as peptic ulcers, osteoporosis, cancer and some viral infections.²

We recently reported the first diastereoselective synthesis of the methyl ester of the scopadulan diterpene (–)-methylthyrsiflorin A (11).³ During this synthesis, we prepared several intermediates whose structures were confirmed by both NMR spectroscopy and physical data. Among these scopadulan precursors, we noted that several tricyclic intermediates possessing a podocarpene skeleton can exist in two different conformations. A complete conformational study using their ¹H NMR data together with molecular mechanics calculations was published separately.⁴

In this paper, we report the ¹H and ¹³C NMR chemical shift assignments obtained from one- and two-dimensional NMR techniques for synthetic **11** and the rest of the scopadulan intermediates (**1–10**) of our synthesis. The ¹³C NMR data for synthetic **11** were identical with those previously reported, whereas the ¹H NMR, HMQC and 1D NOE difference data showed that the assignments of protons H-17 and H-20 were reversed in the literature.^{1d}

RESULTS AND DISCUSSION

The structures and numbering system for compounds **1–11** are presented in Fig. 1. Assignments of ¹H and ¹³C NMR chemical shifts for **1–11** are listed in Tables 1 and 3, respectively, and the multiplicity and coupling constants of some characteristic ¹H NMR signals are shown in Table 2. The obvious signal assignments were made from ¹H and ¹³C NMR and DEPT spectra according to their chemical shifts and multiplicities. The remaining signals were assigned with the aid of double resonance experiments, one-bond heteronuclear (¹H–¹³C) multiple quantum correlation (HMQC) spectra, and some 1D NOE difference experiments (NOED). When it is indicated, the *α* and *β* orientations of protons H-6, H-7, H-11 and H-14 were determined unequivocally by NOED experiments.

From the ¹H NMR data, it is interesting to note the chemical shift of proton H-5 in 4-6 (0.65–0.77 ppm). The high shielding of

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Contract/grant sponsor: DGICYT; Contract/grant number: PB95-1088.



Figure 1. Structures and numbering of the compounds investigated.

this proton may be attributed to the magnetic anisotropy induced by the cyclopropane ring between C-8 and C-9.

From the ¹³C NMR data, the γ -effect observed between C-1 and C-16 in **7–11** is also of interest. This effect is transmitted through proton H-1 α , so the distance between this proton and C-16 determines the intensity of the effect, the greatest effect being with the shortest distance. Therefore, the signal due to C-1 in **7–11** is upfield (about 3–5 ppm to lower frequency) with respect to **1–6** with longer distances between H-1 α and C-16, where the latter exists. The ¹³C NMR spectral data for synthetic **11** were found to be identical with those reported for the methyl ester of the natural (–)methylthyrsiflorin A,^{1d} whereas the HMQC spectrum and NOED experiments indicated that the assignments of protons H-17 and H-20 were originally reversed. In particular, the NOE enhancement of the signal located at 0.92 ppm when proton H-13 was irradiated proved that this signal was due to protons H-17.

Using our synthetic route for the synthesis of **11**, we prepared several intermediates with no precedents in the literature, in particular **4**–**6** which have a novel pentacyclic carbon framework. We think that the ¹H and ¹³C NMR data of these compounds will be useful as reference data for the assignment and characterization of similar compounds.

EXPERIMENTAL

Compounds

All the compounds were prepared as reported previously.⁴

Spectra

¹Ĥ, ¹³C NMR and DEPT spectra were measured on a Varian XL-300 spectrometer (299.95 MHz for ¹H and 75.43 MHz for ¹³C) operating at a probe temperature of 298 K using a dual ¹H/¹³C 5 mm probe. The ¹H measurement conditions were spectral width 4000 Hz, 90° pulse with 18 μ s, acquisition time 3.7 s, number of transients 16–64 and 0.1 Hz digital resolution.







Spectral Assignments and Reference Data

Table 1.	¹ H NMR	chemical	shifts ((δ,	ppm)	for	compounds	1-	-11
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Proton	1	2	3	4	5	6	7	8	9	10	11 ^a
1α	1.19	~1.19	c	1.00	1.05	1.14	1.33	4 50	4.45	4 90 4 50	4.46
1β	1.78	1.76	1.80	1.69	1.77	1.82	1.52	1.50	~1.47	1.30-1.50	~ 1.40
2α	~ 1.45	1.40 - 1.50	c	1.00 1.50	1.20 - 1.28	1.52		1 = 0		104 144	4 40 4 40
2β	1.58	1.58	1.55	1.38-1.50	1.54	1.56	1.51	1.50	$\sim 1.30 - 1.55$	1.36-1.64	1.40 - 1.60
3α	~ 1.15	1.16	c	1.06	1.12	1.15	1.16	1.18	1.14	1.09	~ 1.08
3β	~ 1.40	1.35 - 1.45	c	1.38	1.40	1.43	1.47	1.46	~ 1.40	$\sim \! 1.37$	~ 1.36
5	1.26	1.23	~ 1.25	0.65	0.70	0.77	1.14	1.26	1.00	0.90	0.88
6α	~ 1.65	~ 1.68	c	1.05 1.40	$1.20 - 1.28^{b}$	107 10/	1.69	1.64 ^b	$\sim 1.56^{b}$	$\sim \! 1.51^{b}$	~ 1.48
6β	1.42	1.38 - 1.46	c	1.25-1.40	$1.32 - 1.44^{b}$	1.27-1.36	1.55	$\sim \! 1.55^{\mathrm{b}}$	$\sim 1.32^{b}$	$\sim \! 1.20^{b}$	1.30
7α	4 00 4 05	101 101	c	1 (= 1 00	1 05 1 55		2.39	2.48	1.60 ^b	$\sim \! 1.50^{b}$	$\sim 1.50^{b}$
7β	1.80-1.95	1.84-1.94	c	1.65-1.83	1.35-1.75	1.64-1.76	2.64	2.59	$\sim 1.26^{b}$	$\sim \! 1.26^{b}$	$\sim 1.13^{b}$
8	_	_	_	_	—	_	_	_	2.12	1.80	1.84
11α	2.81	2.66	2.76	1.68	1.41	1.65 ^b	1.94	1.54	1.70 ^b	1.46 ^b	1.50 ^b
11β	2.02	2.16	~ 1.90	2.11	1.78	1.79 ^b	2.30	1.63	1.39 ^b	0.95 ^b	1.03 ^b
13	_	_	_	_	—	_	_	_	_	3.38	4.66
14α	2.02	2.10	2.00	2.04 ^b	1.92 ^b	2.45	5.76	5.74	1.96	$\sim 0.91^{b}$	~ 1.00
14β	2.15	2.10	2.10	2.28 ^b	1.99 ^b		_	_	2.23	1.82 ^b	1.90
15	_	_	_	_	1.36	4 55	_	1.64	$\sim \! 1.56$	~ 1.19	~ 1.18
15′	_	_	_	_	1.96	1.77	_	1.71	~ 1.74	$\sim \! 1.73$	~ 1.73
16	3.68	_	5.78	1.80	0.98	1.22	2.22	2.19	2.02	$\sim \! 1.38$	~ 1.41
16'	_	_	_	_	—	_	2.61	1.44	~ 1.72	$\sim \! 1.73$	~ 1.73
17	1.21	1.25	1.16	0.90	0.88	1.02	1.20	1.18	1.05	1.00	0.92
18	0.87	0.87	0.88	0.79	0.80	0.83	0.92	0.92	0.85	0.82	0.81
19	0.83	0.83	0.84	0.81	0.81	0.83	0.91	0.90	0.82	0.80	0.80
20	0.93	0.94	0.94	1.07	1.05	1.08	1.00	0.95	1.00	0.97	0.96
Ketal	3.75-4.00	4.07	3.80-4.00	3.68-4.00	3.75-3.92	_	_	_	_	_	_

 $^{\rm a}$ Additional side-chain ester $^1{\rm H}$ signals in 11 at δ 3.74 (s) and 3.35 (s).

^b α and β may be interchanged.

^c Signal not assigned.

Table 2. Selected ¹H NMR signals for compounds 1–11: multiplicity and coupling constants (Hz)^a

Proton	1	2	3	4	5	6	7	8	9	10	11
H-5	dd; 11, 2	dd; 11, 2	dd; 11, 2	dd; 12, 4	dd; 9, 6	dd; 11, 4	dd; 13, 3	dd; 13, 3	dd; 12, 3	dd; 12, 3	dd; 12, 3
Η-7α	m	m	m	m	m	m	dddd;	dddd;	m	m	m
							17.5, 13, 6.5, 2.5	17.5, 12, 7, 2			
H-7 β	m	m	m	m	m	m	ddd;	dddd;	m	m	m
							17.5, 5, 1.5	17.5, 6, 2.5, 0.5			
H - 11α	br d; 16.5	br d; 16.5	br d; 16.5	d; 12	d; 12	d ^b ; 12	d; 12	d; 11.5	d ^b ; 12	d ^b ;11.5	d ^b ; 11.5
H-11β	d; 16.5	d; 16.5	d; 16.5	d; 12	d; 12	dd ^b ; 12, 2	dd; 12, 3.5	br d; 11.5	br d ^b ; 12	br d ^b ; 11.5	br d ^b ; 11.5
H-13	—	—	—			—	—	—	—	dd;	ddd;
										10.5, 5.5	11, 6, 1
H-14α	d; 16.5	br s	d; 16.5	d; 15	d; 15	S	d; 2.5	dd; 2, 0.5	dd; 15, 12	m	m
H-14 β	br d; 16.5	br s	br d; 16.5	d; 15	d; 15	S	—	—	dd; 15, 5.5	m	ddd;
											12.5, 6, 6
H-15	—	—	—		dd; 12, 3	m	—	m	m	m	m
H-15′	—	—	—		d; 12	m	—	m	m	m	m
H-16	—	—	—	s	d; 3	d; 3	dd; 17.5, 3.5	ddd; 12.5, 10, 6	m	m	m
H-16′	—	—	—	—	—	—	d; 17.5	m	m	m	m

^{*a*} Some coupling constants were determined by analysis of the multiplet in an NOE difference spectrum and double resonance experiments. ^{*b*} α and β may be interchanged.



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Carbon	1	2	3 ^a	4	5	6	7	8	9	10	11 ^b
1	35.94	35.91	36.28	36.56	36.84	37.05	33.03	32.79	32.55	32.57	32.49
2	18.71	18.71	18.99	18.41	18.79	18.71	18.23	18.40	18.70	18.79	18.72
3	41.55	41.55	41.89	41.59	41.98	41.87	41.64	41.91	42.12	42.20	42.16
4	33.17	33.25	33.36	33.05	33.07	33.08	33.47	33.43	33.13	33.12	33.10
5	51.07	51.12	51.31	47.48	48.16	48.11	47.95	47.22	47.98	47.97	47.94
6	18.49	18.51	19.14	17.75	18.26	18.04	20.62	20.97	21.70	21.83	21.77
7	31.15	31.26	31.67	29.19	30.53	29.53	31.03	30.80	30.29	30.14	29.94
8	122.91	122.41	122.54	32.20	19.68	19.97	171.42	171.12	40.01	37.39	37.14
9	135.75	135.93	136.96	52.29	43.84	42.05	52.57	58.94	52.44	52.47	52.49
10	37.28	37.44	37.73	34.06	33.63	33.56	37.33	37.46	38.83	38.52	38.52
11	33.40	33.56	33.32	31.89	35.30	36.40	42.83	44.12	45.38	44.50	44.47
12	48.41	47.64	52.30	47.89	41.69	49.79	63.69	50.39	53.10	44.11	43.05
13	110.13	110.75	110.85	112.24	111.34	213.36	192.26	204.50	214.64	76.38	80.00
14	39.31	37.33	39.67	41.82	42.56	41.67	125.70	124.37	43.68	38.17	34.19
15	—	_	—	210.99	36.14	36.92	211.82	35.29	36.88	30.69	31.94
16	—	_	—	39.47	26.51	26.49	41.33	29.27	24.09	24.43	24.36
17	18.73	19.02	19.28	10.97	17.72	17.32	14.46	20.22	19.81	23.30	23.11
18	33.05	33.11	33.28	33.71	33.91	33.87	33.71	33.76	33.61	33.62	33.60
19	21.51	21.58	21.77	21.37	21.50	21.46	22.69	22.66	21.96	22.05	22.02
20	18.93	19.11	19.36	17.53	19.76	19.86	19.10	20.86	17.24	17.34	17.31
R	175.07; 51.88	176.13	196.10; 53.52	_	_		_	_	_	_	_
Ketal	64.98; 64.90	64.59; 64.50	64.66; 64.66	65.24; 64.94	64.72; 64.62	_	_	_	—	—	—

Table 3. ¹³C NMR chemical shifts (δ, ppm) for compounds 1–11

 a In C₆D₆.

^b Additional side-chain ester ¹³C signals in **11** at δ 167.17 (s), 166.21 (s), 52.34 (q) and 41.70 (t).

¹H-¹³C HMQC and NOED spectra were measured on a Varian 400 spectrometer (399.95 MHz for ¹H and 100 MHz for ¹³C) equipped with a 5 mm indirect detection probe operating at 298 K. Sample concentrations were typically in the range 5-20 mg per 0.5 ml of CDCl₃. The signal of TMS was taken as the reference. All these experiments were performed either using standard pulse sequences supplied by the spectrometer manufacturer or slightly modified pulse sequences. NOED experiments were typically acquired with 8 K data points covering a spectral width of 3200 Hz and with a 1.5 s presaturation time. Spectra at each presaturation position were interleaved in groups of four scans to minimize artefacts due to instrument inconsistencies and processed with a 1 Hz exponential line broadening to reduce subtraction artefacts. The HMQC spectrum was obtained using a spectral width of 3200 Hz in the ¹H dimension and 16 000 Hz in the ¹³C dimension. A total of 256 increments were collected with eight transients per increment and an acquisition time of 0.1 s.

Acknowledgements

This research was supported by DGICYT (Grant PB95-1088). Miguel A. González is grateful to the Conselleria d'Educació i Ciència de la Generalitat Valenciana for a research fellowship.

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